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Phosphorus Partitions between Carbon-Saturated Iron Melts and CaO-CaF, -SiO, -MnO Slags

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## 1. Introduction

Dephosphorization is one of the most important reactions in steelmaking process. Recently some thermodynamic investigations on the dephosphoriza-Recently some tion of hot metal have been carried out by using soda or lime based slags (1,2). However, so far no thermodynamic investigations have been reported on the dephosphorization of hot metal containing a large amount of manganese by soda or lime based fluxes.

The purpose of this study is to investigate the effects of manganese oxide, slag compositions and temperature on the equilibrium distribution of phosphorus between the molten  $CaO-CaF_2-SiO_2-MnO$ system and carbon-saturated iron with high manganese content under reducing conditions.

## 2. Experimental methods

Two grams of carbon-saturated iron which contains 0.023% phosphorus and 4.60% manganese were equilibrated with two grams of containing MnO 1.0-5.0%,  $\mathrm{SiO}_2$  11-13%,  $\mathrm{CaF}_2$  29-31%,  $\mathrm{CaO}$  48-55% in a graphite crucible under carbon monoxide atmospheres. The temperatures were 1250, 1280, 1300, 1350°C and the experimental time was 20h. The final compositions of slags and metals were determined by ordinary chemical analysis methods.

## 3. Experimental results

Fig.1 shows the effect of manganese oxide content on phosphorus partitions between slag and carbon-saturated iron. The phosphorus distribution ratio slightly decreases with increasing the manganese oxide content in slag. The phosphate capacities defined as

$$c_{P0}_{4}^{3-} = \frac{(\%P0_{4}^{3-})}{P_{P}^{1/2} \cdot P_{0}_{2}^{5/4}}$$

were calculated to be between  $2.5 \times 10^{25}$  and  $3.0 \times 10^{25}$ from the results in Fig.1.

Fig.2 shows the temperature sphorus distribution ratio. dependency of phosphorus The heat reaction:

$$3/2(Ca0) + P + 5/40_2 = 1/2(Ca_3(P0_4)_2) - \cdots (2)$$

calculated from the plots in Fig.2 is -130kcal/mol which is in good agreement with that of Muraki et al (3).

References

1. F.Tsukihashi, F.Matsumoto and N.Sano: Tetsu-to-

Hagane, 70 (1984), No.4, S 203

Hagane, 69 (1983), No.4, S 175 2. K.Ito and N.Sano: Tetsu-to-Hagane, 69 (1983), No.15, p. 1747 3. M.Muraki, H.Fukushima and N.Sano: Tetsu-to-

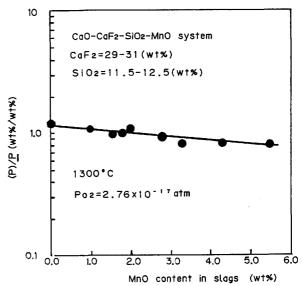


Fig. 1. Influence of MnO content on phosphorus distribution ratio at 1300°C for the  ${\tt CaO-CaF}_2{\tt -SiO}_2{\tt -MnO}$  system.

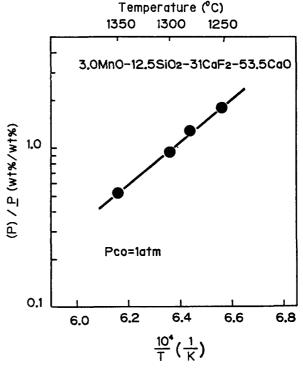


Fig. 2. Temperature dependency of phosphorus distribution between 1250°C and 1350°C for  $3\%\text{Mn}0\text{-}12.5\%\text{Si0}_2\text{-}31\%\text{CaF}_2\text{-}53.5\%\text{CaO}$ .